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N–H Insertion reactions of rhodium carbenoids. Part 5: A convenient route to 1,3-azoles^{\Leftrightarrow}

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Abstract—Dirhodium(II) carboxylate catalysed reaction of diazocarbonyl compounds 2 in the presence of primary amides 1 results in the formation of α -acylaminoketones 3 (12 examples) by N–H insertion reaction of the intermediate rhodium carbene. The 1,4-dicarbonyl compounds 3 are readily converted into structurally diverse oxazoles 4 (11 examples) by cyclodehydration, thiazoles 5 (10 examples) by treatment with Lawesson's reagent, or imidazoles 6 (2 examples) by reaction with ammonia or methylamine. © 2004 Elsevier Ltd. All rights reserved.

1. Introduction

The 1,3-azoles—oxazoles, thiazoles, imidazoles—have attracted the attention of chemists for many years. The heteroaromatic imidazole ring plays a key role in the chemistry of the proteinogenic amino acid histidine, and oxazoles and thiazoles occur widely in a range of natural products, particularly the non-ribosomal peptides.^{2,3} Recently there has been considerable interest in the use of 1,3-azoles as peptide mimetics.^{4–6} The structural diversity of complex naturally occurring 1,3-azoles and the biological activity of synthetic analogues has ensured that new methods continue to be developed for their synthesis.⁷

Of the intermediates available for the synthesis of fivemembered heteroaromatic rings, 1,4-dicarbonyl compounds are among the most versatile. In the field of 1,3-azole synthesis, the cyclodehydration of such a 1,4-dicarbonyl compound (an α -acylaminoketone) is the basis of the Robinson–Gabriel oxazole synthesis.⁷ Although this reaction was discovered some time ago, it continues to undergo modification, for example, the preparation of the intermediate α -acylaminoketone by acylation of α -amino- β -ketoesters,^{4,8} or by oxidation of β -hydroxyamides.⁹ Recently we reported a new variation on the Robinson– Gabriel synthesis in which the key 1,4-dicarbonyl intermediate was obtained by a rhodium carbene N–H insertion reaction,¹⁰ developed specifically for the synthesis of the

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amino acid derived oxazole building blocks of the natural products nostocyclamide and promothiocin A.^{11,12} Subsequently, others used our protocol for the synthesis of oxazole-containing peptide mimetics,⁵ whilst Janda and co-workers have developed a solid-phase variant of this reaction, and applied it to the synthesis of an array of oxazoles.^{13,14} We now report further developments in the use of rhodium carbene N–H insertion reactions and their use in a general approach to the synthesis of 1,3-azoles.

2. Results and discussion

A range of primary amides was selected, comprising formamide 1a, a simple alkanamide 1b, aromatic and heteroaromatic amides 1c-1e, the amide 1f derived from piperidine-4-carboxylic acid (isonipecotic acid), and the oxazole amide 1g. Amide 1d was readily prepared from 5-methoxysalicylic acid, 1f from piperidine-4-carboxamide, and 1g from the corresponding ester 4c (prepared by the method described herein); the other amides are commercially available. A range of six diazocarbonyl compounds 2 was also selected for study. The α -diazo- β ketoesters 2a-2e were prepared by diazo-transfer reaction¹⁵ to the corresponding β -ketoesters using 4-acetamidobenzenesulfonyl azide as the reagent,¹⁶ and azibenzil **2f** was obtained by the literature procedure by oxidation of benzil monohydrazone with manganese(IV) oxide.¹⁷ The carbene N-H insertion reactions were generally carried out using dirhodium tetraacetate as catalyst and 1,2-dichloroethane as solvent, the diazocarbonyl compound being added by syringe pump over about 16 h. The resulting α -acylamino ketones 3 were formed in varying yield (13-82%) (Table 1) with no significant by-products being identified.

[☆] See Ref. 1.

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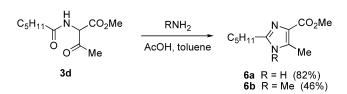
Table 1. Dirhodium(II) catalysed reactions of diazocarbonyl compounds 2 with amides 1, and subsequent cyclisation of the ketoamides 3 to oxazoles 4 and thaizoles 5

	1 2			R^2 N R^4 $-$ 0 R^5 $-$	$X = O$ $Ph_{3}P, I_{2}, Et_{3}N$ $X = S$ Lawesson's reagent		X^ `R⁵	
				3			4 X = O 5 X = S	
Amide 1	\mathbb{R}^2	Diazo 2	\mathbb{R}^4	R ⁵	3-5	3 Yield (%)	4 Yield (%)	5 Yield (%)
1a	Н	2a	CO ₂ Me	Me	а	43	45	60
1a	Н	2f	Ph	Ph	b	54	78	94
1a	Н	2a	CO_2Me	$4-Cl-C_6H_4$	с	55 ^a	65	_
1b (C ₅ H ₁₁	2a	CO ₂ Me	Me	d	82 ^b	79	89
1c	Ph	2b	CO ₂ Et	Me	e	62	80	53
1d :	2-BnO-5-MeO-C ₆ H ₃	2a	CO ₂ Me	Me	f	26	23	55
1d :	2-BnO-5-MeO-C ₆ H ₃	2c	CO ₂ Et	Ph	g	13	67	_
1e :	2-Thienyl	2a	CO ₂ Me	Me	ĥ	80	80	69
1e :	2-Thienyl	2d	CO_2Me	$4-Cl-C_6H_4$	i	36	72	89
1e :	2-Thienyl	2e	CO_2Me	$4-MeO_2C-C_6H_4$	j	74	54	34
1f	N-Boc-piperidin-4-yl	2a	CO_2Me	Me	k	68	66	74
1g :	5-(4-Cl-C ₆ H ₄)oxazol-4-yl	2a	CO ₂ Me	Me	1	65	_	40

^a Dirhodium tetraoctanoate as catalyst in dichloromethane solvent.

^b Dichloromethane solvent.

With a range of 1,4-dicarbonyl compounds **3** in hand, their conversion into 1,3-azoles was investigated. First, cyclodehydration, using the triphenylphosphine-iodine-triethylamine protocol developed by Wipf and Miller,⁹ gave the corresponding oxazoles **4** in 23–80% yield (Table 1), oxazole **4f** being the protected form of the terminal oxazolecarboxylate in the linear lipopeptide amamistatin A.¹⁸ The thiazoles **5** were readily formed from the 1,4dicarbonyl compounds by thionation with Lawesson's reagent.^{19,20} Thus simply heating the α -acylaminoketones **3** with Lawesson's reagent in THF gave the thiazoles **5** in 34–94% yield (Table 1). Finally, two examples of imidazole formation were studied: simply treating 1,4dicarbonyl compound **3d** with ammonium acetate and methylamine gave the imidazoles **6a** and **6b** in 82 and 46% yield, respectively (Scheme 1).²⁰





Thus by extending the scope of the rhodium carbene N–H insertion reaction, a number of α -acylaminoketones has been obtained. These 1,4-dicarbonyl compounds are useful precursors to a range of structurally diverse 1,3-azoles.

3. Experimental

3.1. General

Commercially available reagents were used throughout without further purification unless otherwise stated; solvents were dried by standard procedures. Light petroleum refers to the fraction with bp 40-60 °C and ether refers to diethyl

ether. Reactions were routinely carried out under a nitrogen atmosphere. Analytical thin layer chromatography was carried out using aluminium-backed plates coated with Merck Kieselgel 60 GF₂₅₄. Plates were visualised under UV light (at 254 and /or 360 nm). Flash chromatography was carried out using Merck Kieselgel 60 H silica or Matrex silica 60. Fully characterised compounds were chromatographically homogeneous. IR spectra were recorded in the range 4000–600 cm⁻¹ using a Nicolet Magna FT-550 spectrometer. ¹H and ¹³C NMR spectra were recorded using Bruker 300 and 400 MHz instruments (¹H frequencies, corresponding ¹³C frequencies are 75 and 100 MHz); J values were recorded in Hz. In the ¹³C NMR spectra, signals corresponding to CH, CH₂ or Me groups are noted; all others are C. High and low-resolution mass spectra were recorded on a Micromass GCT TOF High Resolution Mass Spectrometer, or at the EPSRC Mass Spectrometry Service (Swansea).

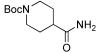


3.1.1. 2-Benzyloxy-5-methoxybenzamide 1d. (a) To a solution of 5-methoxysalicylic acid (2.00 g, 12 mmol) and potassium carbonate (8.20 g, 59 mmol) in DMF (40 ml) cooled to 0 °C was added benzyl bromide (4.24 ml, 36 mmol). The mixture was then stirred at 0 °C for 30 min and then overnight at ambient temperature. Ethyl acetate (100 ml) was added to the mixture and washed with aqueous potassium hydrogen sulfate (1 M; 100 ml), water (100 ml), saturated potassium hydrogen carbonate solution (100 ml) and saturated brine (100 ml). The organic layer was dried (MgSO₄) and concentrated to give a dark oil that was then dissolved in methanol (38 ml) and sodium hydroxide solution (40%, 12 ml) and stirred at ambient for 1 h. The reaction mixture was then acidified to pH 1 with dilute

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hydrochloric acid solution (30 ml) and extracted with diethyl ether (3×15 ml). The organic extracts were dried (MgSO₄) and concentrated to yield the crude compound which was recrystallised from hexane–ethyl acetate (3:2) to yield 2-benzyloxy-5-methoxybenzoic acid as a colourless crystalline solid (1.70 g, 63%); mp 95–96 °C (lit.¹⁸ mp 85–87 °C); ν_{max} (KBr)/cm⁻¹ 3440, 3001, 2962, 2929, 2910, 2835, 1695, 1597, 1502, 1454, 1416, 1389, 1329, 1296, 1221, 1047, 1018, 914, 874, 854, 822, 744, 700; $\delta_{\rm H}$ (300 MHz; CDCl₃) 9.70 (1H, s br, OH), 7.70 (1H, m, ArH), 7.43 (5H, m, ArH), 7.09 (2H, m, ArH), 5.26 (2H, s, OCH₂Ph), 3.83 (3H, s, OMe); $\delta_{\rm C}$ (75 MHz; CDCl₃) 165.5, 155.0, 151.9, 134.8, 129.6 (CH), 129.5 (CH), 128.4 (CH), 122.5 (CH), 119.0, 116.6 (CH), 115.3 (CH), 73.4 (CH₂), 56.3 (Me).

(b) To a solution of the above acid (1.50 g, 58 mmol) and triethylamine (0.81 ml, 58 mmol) in THF (35 ml) cooled to 0 °C was added ethyl chloroformate (0.56 ml, 58 mmol) and stirred for 15 min. Ammonia solution (30%, 35 ml) in THF (15 ml) was then added to the reaction and stirred for 15 min and then concentrated. The solid residue was then partition with dichloromethane (30 ml) and water (30 ml). The aqueous was then washed again with dichloromethane (30 ml) and the combined organics were washed with saturated sodium hydrogen carbonate solution (30 ml), brine (30 ml), dried (MgSO₄) and concentrated to yield the title compound as a light brown crystalline solid (1.33g, 89%); mp 121-123 °C; (Found: M⁺, 257.1064. C₁₅H₁₅NO₃ requires 257.1052); ν_{max} (KBr)/cm⁻¹ 3452, 2426, 3314, 3252, 3160, 2924, 2827, 1660, 1598, 1578, 1491, 1429, 1368; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.82 (2H, m, ArH), 7.40 (4H, m, ArH), 7.01 (2H, m, ArH), 5.84 (2H, s br, NH₂), 5.14 (2H, s, OCH₂Ph), 3.83 (3H, s, OMe); m/z (CI) 257 (M⁺, 71%), 241 (22), 215 (10), 195 (7), 181 (3), 151 (16), 137 (7), 119 (3), 91 (54).



3.1.2. 1-tert-Butoxycarboxypiperidine-4-carboxamide 1f. To a stirred solution of the piperidine-4-carboxamide (1.00 g, 7.8 mmol) and di-tert-butyl dicarbonate (2.20 g, 10.1 mmol) in acetonitrile (15 ml) was added DMAP (95 mg, 0.78 mmol). The resulting solution was stirred overnight and then the solvent was removed under reduced pressure. The solid residue was then dissolved in dichloromethane (40 ml) and washed with saturated sodium hydrogen carbonate solution (40 ml). The aqueous was then extracted with dichloromethane (40 ml) and then the combined organic extracts were washed with saturated ammonium chloride solution (60 ml), water (60 ml) and brine (60 ml) and then dried (MgSO₄) and concentrated to yield the desired product as a colourless solid (1.70 g, 92%); mp 152–155 °C (lit.²¹ mp 154–156 °C); ν_{max} (KBr)/cm⁻¹ 3363, 3190, 2976, 2935, 2860, 1687, 1660, 1632, 1479, 1435, 1365, 1288, 1234, 1180, 1146, 1119, 1034, 926, 872, 769; $\delta_{\rm H}$ (300 MHz; CDCl₃) 5.79 (1H, s, NH), 5.64 (1H, s, NH), 4.11 (2H, m, 2×CH), 2.72 (2H, m, 2×CH), 2.28 (1H, m, CH), 1.64 (2H, m, 2×CH), 1.57 (2H, m, 2×CH), 1.44 (9H, s, CMe₃); δ_C (75 MHz; CDCl₃) 177.4, 155.0, 80.1, 43.6 (CH), 29.0 (CH₂), 43.0 (CH₂), 28.8 (Me); *m/z* (CI) 229 (MH⁺, 15%), 217 (4), 201 (14), 184 (2), 174 (6), 173 (100), 156 (7), 155 (76), 129 (6), 112 (7), 106 (2).



3.1.3. 5-(4-Chlorophenyl)-oxazole-4-carboxamide 1g. (a) A solution of oxazole ester 4c (500 mg, 2 mmol) and sodium hydroxide (421 mg, 10 mmol) in THF (25 ml) and water (8 ml) was stirred overnight. The reaction mixture was concentrated and the residue was partitioned between dichloromethane (80 ml) and water (80 ml). The aqueous layer was then acidified to pH 1 with diluted hydrochloric acid and extracted with dichloromethane (150 ml). The organic layer was then washed with brine (75 ml), dried (MgSO₄) and the solvent removed under reduced pressure to 5-(4-chlorophenyl)oxazole-4-carboxylic acid as a colourless crystalline solid (411 mg, 87%); mp 178-179 °C; (Found: C, 53.5; H, 2.3; N, 6.1. C₁₀H₆ClNO₃ requires C, 53.7; H, 2.7; N, 6.3%); (Found: M⁺, 224.0131. $C_{10}H_6^{35}$ ClNO₃ requires 224.0114); ν_{max} (KBr)/cm⁻¹ 3128, 3035, 2956, 2924, 2854, 1722, 1701, 1585, 1535, 1491, 1294, 1273, 1234, 1126, 1099, 1068, 1016, 989, 951, 883, 793, 762, 729, 640; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.10 (2H, d, J=8.7 Hz, ArH), 7.93 (1H, s, H-2), 7.41 (2H, d, J=8.7 Hz, ArH), 5.66 (1H, br s, OH); δ_{C} (75 MHz; CDCl₃) 163.2, 153.1, 151.3 (CH), 135.3, 130.4 (CH), 129.0 (CH), 127.6, 126.0; m/z (CI) 226/224 (M⁺, 7/13%), 209 (4), 208/206 (33/ 100), 182/180 (5/15), 179 (7), 154/152 (3/7), 139 (11), 125 (5).

(b) To a solution of the above acid (480 mg, 2 mmol) and triethylamine (0.30 ml, 2 mmol) in THF (30 ml) cooled to 0 °C was added ethyl chloroformate (0.21 ml, 2 mmol), and the mixture stirred for 15 min at 0 °C. Aqueous ammonia (30% w/w, 20 ml) and THF (15 ml) were then added, and the reaction mixture was stirred at ambient for 15 min. The reaction mixture was then concentrated under reduced pressure and the residue partitioned between water (100 ml) and dichloromethane (100 ml). The aqueous layer was extracted with further dichloromethane (100 ml) and the combined organics were washed with saturated sodium hydrogen carbonate solution (100 ml) and brine (100 ml), dried (Mg₂SO₄) and the solvent removed under reduced pressure to yield the title compound as a colourless crystalline solid (275 mg, 58%); mp 245–247 °C; (Found: M⁺, 222.0194. $C_{10}H_7^{35}ClN_2O_2$ requires 222.0196); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 2404, 3361, 3290, 3201, 3128, 3080, 2924, 2852, 1699, 1608, 1597, 1533, 1491, 1400, 1329, 1257, 1194, 1119, 1093, 1061, 1016, 987, 953, 835, 793, 742, 708, 667; δ_H (300 MHz; CDCl₃) 8.51 (1H, s, H-2), 8.18 (2H, d, J=8.7 Hz, ArH), 7.72 (1H, s, NH), 7.60 (1H, s, NH), 7.51 (2H, d, J=8.7 Hz, ArH); δ_C (75 MHz; DMSO) 163.0, 150.6 (CH), 150.5, 150.4, 134.8, 129.9 (CH), 128.9 (CH), 126.2; *m/z* (CI) 224/222 (M⁺, 11/17%), 221 (5), 208/206 (34/100), 187 (4), 186 (12).

3.2. General method for diazo transfer

To a solution of the β -ketoester substrate (10 mmol) and

4-acetamidobenzenesulfonyl azide¹⁶ (11 mmol) in acetonitrile (60 ml) at 0 °C was added triethylamine (30 mmol) dropwise. After stirring at room temperature for 16 h the reaction mixture was concentrated in vacuo and the resultant solid was triturated with ether–light petroleum. The filtrate was concentrated in vacuo and purified by flash chromatography on silica gel eluting with ethyl acetate– light petroleum (1:4) to yield the desired product.

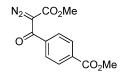
3.2.1. Methyl 2-diazo-3-oxobutanoate 2a. Obtained as a yellow oil according to the general procedure in 92% yield; data as previously described.¹⁰

3.2.2. Ethyl 2-diazo-3-oxobutanoate 2b. According to the general procedure the title compound was obtained as a yellow oil (83%) (lit.²² data not given); (Found: M⁺, 156.0531. C₆H₈N₂O₃ requires 156.0535); ν_{max} (film)/cm⁻¹ 2985, 2939, 2912, 2877, 2141, 1716, 1660, 1595, 1533, 1458, 1373, 1319, 1251, 1155, 1074, 1022, 966, 858, 744, 639; $\delta_{\rm H}$ (300 MHz; CDCl₃) 4.24 (2H, q, *J*=7.1 Hz, OCH₂Me), 2.41 (3H, s, Me), 1.27 (3H, t, *J*=7.1 Hz, CH₂Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 188.4, 159.5, 74.5, 59.6 (CH₂), 26.4 (Me), 12.4 (Me); *m*/*z*(CI) 156 (M⁺, 26%), 129 (16), 111 (4), 102 (5), 101 (100), 87 (3), 85 (17), 83 (8).

3.2.3. Ethyl 2-diazo-3-oxo-3-phenylpropanoate 2c. According to the general procedure, the title compound was obtained as a yellow oil (99%) (lit.²² data not given); (Found: C, 60.8; H, 4.8; N, 13.0. $C_{11}H_{10}N_2O_3$ requires C, 60.6; H, 4.6; N, 12.9%); $\nu_{max}(film)/cm^{-1}$ 3058, 2976, 2940, 2904, 2868, 2136, 1721, 1685, 1629, 1598, 1578, 1450, 1368, 1301, 1260, 1178, 1112, 1015, 938, 917, 789, 748, 692, 671; δ_{H} (300 MHz; CDCl₃) 7.62 (2H, m, ArH), 7.52 (1H, m, ArH), 7.40 (2H, m, ArH), 4.24 (2H, q, *J*=7.3 Hz, OCH₂Me), 1.25 (3H, t, *J*=7.3 Hz, CH₂Me); δ_{C} (75 MHz; CDCl₃) 186.9, 160.9, 137.0, 132.2 (CH), 128.4 (CH), 128.0 (CH), 76.1, 61.4 (CH₂), 14.0 (Me); *m/z* (CI) 218 (M⁺, 8%), 193 (65), 175 (21), 163 (100), 145 (85), 105 (53), 91 (4).



3.2.4. Methyl 3-(4-chlorophenyl)-2-diazo-3-oxopropanoate 2d. According to the general procedure the title compound was obtained as a yellow solid (89%); mp 104 °C (chloroform) (lit.²³ mp 105.5–107.5 °C); (Found: C, 50.1; H, 2.7; N, 11.7. C₁₀H₇ClN₂O₃ requires C, 50.3; H, 2.9; N, 11.7%); ν_{max} (KBr)/cm⁻¹ 2955, 2919, 2848, 2141, 1716, 1624, 1583, 1434, 1342, 1265, 1127, 1086, 1015, 968, 902, 840, 753, 733, 687; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.51 (2H, d, $\begin{array}{l} J{=}8.7 \ \text{Hz}, \ \text{ArH}), \ 7.32 \ (2\text{H}, \ \text{d}, \ J{=}8.7 \ \text{Hz}, \ \text{ArH}), \ 3.73 \ (3\text{H}, \ \text{s}, \ \text{OMe}); \ \delta_{\text{C}} \ (75 \ \text{MHz}; \ \text{CDCl}_3) \ 183.7, \ 159.3, \ 136.7, \ 133.2, \ 128.0 \ (\text{CH}), \ 126.3 \ (\text{CH}), \ 74.6, \ 50.5 \ (\text{Me}); \ m/z \ (\text{EI}) \ 240/238 \ (\text{M}^+, \ 10/39\%), \ 212 \ (7), \ 210 \ (15), \ 154 \ (4), \ 152 \ (8), \ 141 \ (67), \ 139 \ (100), \ 123 \ (48), \ 111 \ (78), \ 75 \ (47). \end{array}$



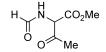
3.2.5. Methyl 2-diazo-3-(4-methoxycarbonylphenyl)-3oxopropanoate 2e. According to the general procedure the title compound was obtained as a yellow solid (99%); mp 70–72 °C (ethyl acetate–light petroleum); (Found: C, 55.1; H, 3.6; N, 10.8. $C_{12}H_{10}N_2O_5$ requires C, 55.0; H, 3.8; N, 10.7%) (Found: M⁺, 262.0588. $C_{12}H_{10}N_2O_5$ requires 262.0590); ν_{max} (KBr)/cm⁻¹ 3027, 2950, 2996, 2919, 2853, 2131, 1721, 1624, 1434, 1409, 1281, 1189, 1132, 1107, 1020, 974, 963, 902, 866, 820, 784, 743, 707, 677; δ_{H} (300 MHz; CDCl₃) 8.02 (2H, d, *J*=8.6 Hz, ArH), 7.59 (2H, d, *J*=8.6 Hz, ArH), 3.87 (3H, s, OMe), 3.72 (3H, s, OMe); δ_C (75 MHz; CDCl₃) 186.8, 166.6, 161.5, 141.2, 133.5, 129.5 (CH), 128.6 (CH), 52.9 (Me), 52.8 (Me), diazo carbon not observed; *m*/*z*(CI) 262 (M⁺, 11%), 235 (20), 205 (6), 204 (10), 203 (100), 191 (16), 179 (4), 163 (28), 159 (11), 131 (4).



3.2.6. Azibenzil 2f. Prepared in 87% yield by oxidation of benzil monohydrazone (2.50 g, 11 mmol) in chloroform (38 ml) using activated manganese dioxide (3.90 g, 44 mmol) according to the literature procedure,¹⁷ mp 78–80 °C (lit.²⁴ mp 79–80 °C).

3.3. General procedure for N–H insertion reactions; preparation of α -acylamino ketones 3

To a solution of the amide 1 (5 mmol) and dirhodium tetraacetate (2.5 mol%) in 1,2-dichloroethane (10 ml), heated to reflux, was added a solution of the diazo compound 2 (7 mmol) in 1,2-dichloroethane dropwise over 16 h. The reaction mixture was then heated for a further 2–4 h until TLC analysis showed that the reaction was complete. The mixture was evaporated in vacuo and purified on silica gel eluting with ethyl acetate–light petroleum (1:4) to yield the product.



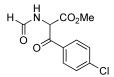
3.3.1. *N*-(**1-Methoxycarbonyl-2-oxopropyl)formamide 3a.** According to the general procedure, using formamide **1a**, diazo compound **2a** and 1,2-dichloromethane as solvent, the title compound was obtained as a colourless crystalline solid (43%); mp 68–70 °C (ethyl acetate–light petroleum); (Found: C, 45.2; H, 5.7; N, 8.7. C₆H₉NO₄ requires C, 45.3; H, 5.7; N, 8.8%); (Found: M⁺, 160.0623. C₆H₉NO₄ requires 160.0610); ν_{max} (KBr)/cm⁻¹ 3319, 3017, 2950, 2925, 2894,

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2853, 1747, 1721, 1639, 1521, 1434, 1388, 1373, 1347, 1250, 1209, 1163, 1107, 1035, 968, 933, 886, 758, 661, 600; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.28 (1H, s, CHO), 7.85 (1H, s br, NH), 5.34 (1H, d, *J*=7.0 Hz, *CH*NH), 3.86 (3H, s, OMe), 2.43 (3H, s, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 198.0, 166.5, 160.8 (CH), 62.0 (CH), 53.9 (Me), 28.4 (Me); *m/z* (CI) 160 (M⁺, 36%), 156 (6), 132 (100), 128 (53), 117 (11), 85 (6), 83 (16).

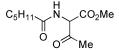


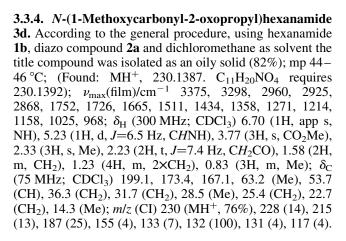
3.3.2. N-(2-Oxo-1,2-diphenylethyl)formamide 3b. To a suspension of formamide 1a (234 mg, 5 mmol) and dirhodium tetraacetate (62 mg, 0.12 mmol) in dichloromethane (20 ml) was added a solution of azibenzil 2f (1.5 g, 6.7 mmol) in dichloromethane (30 ml) over 15 min and stirred at room temperature overnight. The reaction was the heated to reflux for 4 h and then evaporated in vacuo and purified on silica gel eluting with ethyl acetate-light petroleum (2:3) to yield the title product as a beige solid (670 mg, 54%); mp 119–120 °C (lit.²⁵ mp 122 °C); (Found: C, 74.9; H, 5.5; N, 5.7. C₁₅H₁₃NO₂ requires C, 75.3; H, 5.5; N, 5.9%); ν_{max} (KBr)/cm⁻¹ 3370, 3063, 3032, 2919, 2858, 2751, 1690, 1660, 1593, 1578, 1491, 1445, 1383, 1322, 1296, 1255, 1219, 1189, 1066, 984, 933, 881, 851, 779, 758, 738, 692, 677, 656; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.28 (1H, s, CHO), 7.99 (2H, m, ArH), 7.54 (1H, m, NH), 7.42 (4H, m, ArH), 7.32 (4H, m, ArH), 6.66 (1H, d, J=7.4 Hz, CHNH); δ_C (75 MHz; CDCl₃) 195.6, 160.7 (CH), 137.3, 134.5, 134.4 (CH), 129.8 (CH), 129.7 (CH), 129.3 (CH), 129.0 (CH), 128.7 (CH), 57.6 (CH); m/z (CI) 239 (M⁺, 3%), 212 (51), 196 (12), 195 (94), 167 (7), 149 (4), 134 (8), 105 (4).

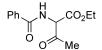


3.3.3. N-[2-(4-Chlorophenyl)-1-methoxycarbonyl-2oxoethyl]formamide 3c. (a) According to the general procedure, using formamide 1a, diazo compound 2d and 1,2-dichloroethane as solvent, the title compound was obtained as a colourless solid (21%); mp 116-119 °C (lit.²⁶ mp 116–118 °C); (Found: C, 51.8; H, 3.7; N, 5.2. C₁₁H₁₀ClNO₄ requires C, 51.7; H, 3.9; N, 5.5%); $\nu_{\rm max}$ (KBr)/cm⁻¹ 3365, 3091, 3076, 3045, 3012, 2958, 2941, 2912, 2877, 2848, 1747, 1699, 1666, 1585, 1570, 1498, 1435, 1406, 1354, 1273, 1252, 1211, 1190, 1167, 1093, 999, 964, 849, 762, 660, 611, 538, 472; δ_H (300 MHz; CDCl₃) 8.23 (1H, s, CHO), 8.00 (2H, d, J=8.7 Hz, ArH), 7.43 (2H, d, J=8.7 Hz, ArH), 6.98 (1H, d, J=7.5 Hz, NH), 6.17 (1H, d, J=7.5 Hz, CHNH), 3.68 (3H, s, OMe); $\delta_{\rm C}$ (75 MHz; CDCl₃) 190.2, 166.9, 160.9 (CH), 141.8, 132.5, 131.4 (CH), 129.7 (CH), 56.8 (CH), 54.0 (Me); m/z (EI) 255/257 (M⁺, 7/1%), 227 (4), 224 (16), 223 (3), 196 (31), 168 (7), 140 (100), 139 (64), 133 (47), 113 (84), 112 (99), 111 (96), 104 (28), 85 (23), 77 (45), 75 (69), 51 (59).

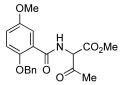
(b) According to the general procedure, using dichloromethane as solvent, the title compound was obtained as a colourless solid (41%). See above for data. (c) According to the general procedure, using dichloromethane as solvent and dirhodium tetraoctanoate as catalyst, the title compound was obtained as a colourless solid (55%). See above for data.





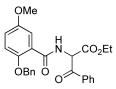


3.3.5. N-(1-Ethoxycarbonyl-2-oxopropyl)benzamide 3e. According to the general procedure using benzamide 1c and diazo compound 2b, the title compound was obtained as a colourless oil (62%) (lit.²⁷ data not given); (Found: M⁺, 250.1080. $C_{13}H_{15}NO_4$ requires 250.1079); $\nu_{max}(film)/cm^{-1}$ 3401, 3058, 3027, 2981, 2940, 2868, 1752, 1726, 1659, 1603, 1578, 1516, 1481, 1445, 1373, 1337, 1265, 1209, 1178, 1102, 1071, 1015, 861, 799, 712, 692; δ_H (300 MHz; CDCl₃) 7.83 (2H, m, ArH), 7.46 (3H, m, ArH), 7.41 (1H, d, J=6.4 Hz, NH), 5.43 (1H, d, J=6.4 Hz, CHNH), 4.29 (2H, q, J=7.1 Hz, OCH₂Me), 2.46 (3H, s, Me), 1.29 (3H, t, J=7.1 Hz, CH₂Me); δ_{C} (75 MHz; CDCl₃) 199.1, 167.2, 166.5, 133.4, 132.5 (CH), 129.0 (CH), 127.6 (CH), 63.9 (CH), 63.1 (CH₂), 20.5 (Me), 14.4 (Me); *m/z* (CI) 250 (M⁺, 88%), 233 (4), 232 (23), 208 (6), 207 (50), 204 (71), 188 (3), 172 (3), 161 (13), 160 (3), 133 (5), 122 (7), 105 (100).

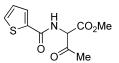


3.3.6. 2-Benzyloxy-5-methoxy-*N***-(1-methoxycarbonyl-2-oxopropyl)benzamide 3f.** According to the general procedure using amide **1d** and diazo compound **2a** the title compound was obtained as a colourless solid (26%); mp 65–67 °C (ethyl acetate–light petroleum) (lit.¹⁸ mp not given); (Found: M⁺, 371.1362. C₂₀H₂₁NO₆ requires 371.1369); ν_{max} (KBr)/cm⁻¹ 3437, 3365, 2924, 2356, 2336, 1752, 1721, 1644, 1603, 1496, 1455, 1440, 1388, 1352, 1306, 1281, 1214, 1173, 1143, 1040, 999, 892, 856, 815, 769, 743, 697, 666, 615; $\delta_{\rm H}$ (300 MHz; CDCl₃) 9.23

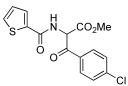
(1H, d, J=6.0 Hz, NH), 7.70 (1H, m, ArH), 7.40 (5H, m, ArH), 6.98 (2H, m, ArH), 5.43 (1H, d, J=6.0 Hz, CHNH), 5.26 (2H, s, OCH_2Ph), 3.80 (3H, s, OMe), 3.74 (3H, s, OMe), 2.36 (3H, s, Me); δ_C (100 MHz; $CDCl_3$) 198.6, 166.6, 164.7, 154.0, 151.4, 135.7, 128.7 (CH), 128.4 (CH), 128.0 (CH), 121.2, 120.2 (CH), 115.3 (CH), 115.0 (CH), 72.1 (CH₂), 63.9 (CH), 55.8 (Me), 53.1 (Me), 27.9 (Me); m/z(EI) 371 (M⁺, 3%), 353 (4), 281 (3), 258 (6), 257 (47), 255 (6), 231 (3), 178 (6), 151 (45), 150 (65), 102 (59), 91 (100), 59 (7).



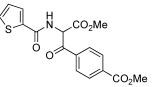
3.3.7. 2-Benzyloxy-N-(1-ethoxycarbonyl-2-oxo-2-phenylethyl)-5-methoxybenzamide 3g. According to the general procedure, using amide 1d and diazo compound 2c, the title compound was obtained as a colourless solid (13%); mp 85-88 °C (diethyl ether); (Found: C, 69.8; H, 5.5; N, 3.0. $C_{26}H_{25}NO_6$ requires C, 69.8; H, 5.6; N, 3.1%); $\nu_{max}(KBr)/$ cm⁻¹ 3334, 3088, 3073, 2996, 2976, 2950, 2930, 2899, 2832, 1737, 1690, 1655, 1614, 1598, 1578, 1506, 1486, 1445, 1281, 1224, 1204, 1184, 1158, 1086, 1045, 1025, 979, 948, 902, 815, 769, 743, 687, 645; δ_H (300 MHz; CDCl₃) 9.66 (1H, d, J=7.1 Hz, NH), 8.34 (2H, d, J=7.7 Hz, ArH), 7.94 (1H, s, H-6), 7.82 (1H, m, ArH), 7.71 (3H, m, ArH), 7.62-7.48 (4H, m, ArH), 7.19 (2H, m, ArH), 6.59 (1H, d, J=7.1 Hz, CHNH), 5.50 (2H, s, OCH₂Ph), 4.36 (2H, q, J=7.0 Hz, CH₂Me), 4.01 (3H, s, OMe), 1.34 (3H, t, J=7.0 Hz, CH₂Me); δ_{C} (75 MHz; CDCl₃) 191.2, 166.4, 164.4, 153.5, 151.0, 135.4, 134.1 (CH), 133.8, 129.1 (CH), 128.4 (CH), 128.3 (CH), 127.9 (CH), 127.4 (CH), 121.1, 119.8 (CH), 114.9 (CH), 114.7 (CH), 71.6 (CH₂), 61.9 (CH₂), 58.8 (CH), 55.4 (Me), 13.5 (Me); *m*/*z* (EI) 447 (M⁺, 22%), 429 (53), 401 (16), 357 (20), 340 (26), 339 (44), 297 (20), 258 (28), 257 (100), 239 (57), 212 (28), 151 (50), 150 (85), 91 (94).



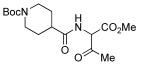
3.3.8. N-(1-Methoxycarbonyl-2-oxopropyl)thiophene-2carboxamide 3h. According to the general procedure, using thiophene-2-carboxamide 1e, diazo compound 2a and 1,2-dichloromethane as solvent, the title compound was obtained as a colourless oil (80%); (Found: C, 49.8; H, 4.5; N, 5.4. C₁₀H₁₁NO₄S requires C, 49.8; H, 4.6; N, 5.8%); $\nu_{\rm max}({\rm film})/{\rm cm}^{-1}$ 3334, 3088, 3073, 2996, 2976, 2950, 2930, 2899, 2832, 1737, 1690, 1655, 1614, 1598, 1578, 1506, 1486, 1445, 1281, 1224, 1204, 1184, 1158, 1086, 1045, 1025, 979, 948, 902, 815, 769, 743, 687, 645; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.61 (1H, dd, J=1.1, 3.8 Hz, ArH), 7.51 (1H, dd, J=1.1, 4.9 Hz, ArH), 7.19 (1H, d, J=6.4 Hz, NH), 7.08 (1H, dd, J=3.8, 4.9 Hz, ArH), 5.40 (1H, d, J=6.4 Hz, CHNH), 3.82 (3H, s, OMe), 2.42 (3H, s, OMe); $\delta_{\rm C}$ (75 MHz; CDCl₃) 198.7, 166.9, 161.7, 137.6, 131.5 (CH), 129.5 (CH), 128.2 (CH), 63.6 (CH), 53.9 (Me), 28.5 (Me); *m*/*z* (EI) 241 (M⁺, 20%), 199 (85), 167 (63), 112 (18), 110 (100), 83 (14).



3.3.9. N-[2-(4-Chlorophenyl)-1-methoxycarbonyl-2oxoethyl)thiophene-2-carboxamide 3i. According to the general procedure, using thiophene-2-carboxamide 1e, diazo compound 2d and 1,2-dichloromethane as solvent, the title compound was obtained as a colourless crystalline solid (36%); mp 100 °C (diethyl ether); (Found: C, 53.2; H, 3.4; N, 4.0. C₁₅H₁₂ClNO₄S requires C, 53.3; H, 3.6; N, 4.2%); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 3421, 3314, 3109, 3088, 3073, 3037, 2991, 2950, 2843, 1737, 1685, 1644, 1619, 1578, 1532, 1501, 1432, 1358, 1312, 1271, 1224, 1199, 1158, 1086, 1004, 958, 927, 861, 840, 764, 712, 605; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.12 (2H, d, J=8.7 Hz, ArH), 7.64 (1H, dd, J=1.1, 3.9 Hz, ArH), 7.54 (1H, dd, J=1.1, 5.0 Hz, ArH), 7.50 (2H, d, J=8.7 Hz, ArH), 7.32 (1H, d, J=7.1 Hz, NH), 7.11 (1H, dd, 3.9, 5.0, ArH), 6.31 (1H, d, J=7.1 Hz, CHNH), 3.75 (3H, s, OMe); δ_{C} (75 MHz; CDCl₃) 190.7, 167.3, 161.8, 141.7, 137.6, 132.8, 131.6 (CH), 131.4 (CH), 129.7 (CH), 129.6 (CH), 128.2 (CH), 58.5 (CH), 53.5 (Me); m/z (CI) 340/338 (M⁺, 32/100%), 322/320 (3/7), 308/306 (13/39), 278 (3), 254 (6), 226 (4), 140 (3), 138 (6).



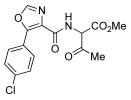
3.3.10. N-[1-Methoxycarbonyl-2-(4-methoxycarbonylphenyl)-2-oxoethyl)thiophene-2-carboxamide 3j. According to the general procedure, using thiophene-2-carboxamide 1e, diazo compound 2e and 1,2-dichloromethane as solvent the title compound was obtained as a colourless crystalline solid (74%); mp 188–190 °C (diethyl ether); (Found: C, 56.3; H, 4.1; N, 3.7. $C_{17}H_{15}NO_6S$ requires C, 56.5; H, 4.2; N, 3.9%); $\nu_{max}(KBr)/cm^{-1}$ 3309, 3249, 3113, 3097, 3067, 3041, 3017, 2957, 2848, 1710, 1690, 1634, 1530, 1502, 1453, 1433, 1421, 1405, 1361, 1317, 1285, 1261, 1237, 1205, 1169, 1113, 1041, 1005, 964, 928, 872; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.22 (2H, d, J=8.7 Hz, ArH), 8.17 (2H, d, J=8.7 Hz, ArH), 7.65 (1H, dd, J=1.1, 3.8 Hz, ArH), 7.54 (1H, dd, J=1.1, 4.9 Hz, ArH), 7.33 (1H, d, J=7.3 Hz, NH), 7.11 (1H, dd, J=3.8, 4.9 Hz, ArH), 6.37 (1H, d, J=7.3 Hz, CHNH), 3.96 (3H, s, OMe), 3.75 (3H, s, OMe); δ_C (75 MHz; CDCl₃) 191.7, 167.1, 166.3, 161.8, 137.7, 137.5, 135.4, 131.6 (CH), 130.4 (CH), 129.9 (CH), 129.7 (CH), 128.2 (CH), 58.8 (CH), 53.9 (Me), 53.0 (Me); m/z (CI) 362 (M⁺, 100%), 344 (7), 330 (25), 278 (3), 176 (3), 163 (8).



3.3.11. 1-tert-Butyloxycarbonyl-N-(1-methoxycarbonyl-2-oxopropyl)piperidine-4-carboxamide 3k. According to

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the general procedure the title product was isolated from reaction of amide 1f and diazo compound 2a in dichloromethane as an oily solid (68%); mp 76-79 °C; (Found: MH⁺, 343.1871. C₁₆H₂₇N₂O₆ requires 343.1869); $\nu_{\rm max}$ (KBr)/cm⁻¹ 3280, 3057, 2978, 2960, 2937, 2860, 1757, 1695, 1639, 1541, 1435, 1365, 1342, 1279, 1250, 1214, 1173, 1107, 968, 764, 661; δ_H (300 MHz; CDCl₃) 6.68 (1H, d, J=6.3 Hz, NH), 5.24 (1H, d, J=6.3 Hz, CHNH), 4.14-4.07 (2H, m, 2×CH), 3.81 (3H, s, OMe), 2.81-2.73 (2H, m, 2×CH), 2.39 (4H, m, Me+CH), 1.85-1.81 (2H, m, 2×CH), 1.70–1.56 (2H, m, 2×CH), 1.44 (9H, s, CMe₃); δ_{C} (75 MHz; CDCl₃) 198.7, 174.5, 166.9, 155.0, 80.1, 63.2 (CH), 53.8 (Me), 43.5 (CH₂), 43.0 (CH), 28.8 (Me), 28.7 (CH₂), 28.5 (Me); *m*/*z* (CI) 343 (MH⁺, 5%), 327 (2), 315 (6), 299 (2), 288 (7), 287 (73), 269 (40), 241 (10), 218 (12), 186 (7), 174 (22), 173 (100), 155 (21), 133 (6), 131 (83), 115 (7), 102 (7), 75 (17).



3.3.12. 5-Chlorophenyl-N-(1-methoxycarbonyl-2-oxopropyl)oxazole-4-carboxamide 3l. To a solution of amide 1g (385 mg, 1.73 mmol) and dirhodium tetraoctanoate (34 mg, 0.04 mmol) in dichloromethane (20 ml) heated to reflux was added a solution of diazo compound 2a (320 mg, 2.25 mmol) in dichloromethane (10 ml) dropwise over 4 h. The reaction mixture was then concentrated under reduced pressure and the residue was triturated with diethyl ether to yield the title compound as a beige solid (280 mg, 48%). The trituration liquors were then reduced in vacuo and purified by flash column chromatography to yield further title compound (101 mg, 17%, total 381 mg, 65%); mp 122-125 °C; (Found: M⁺, 337.0599. C₁₅H₁₃³⁵ClN₂O₅ requires 337.0591); ν_{max} (KBr)/cm⁻¹ 3401, 3334, 3135, 3083, 3032, 2960, 2930, 2853, 1747, 1660, 1588, 1527, 1496, 1440, 1363, 1332, 1260, 1224, 1163, 1091, 1055, 1009, 979, 943, 927, 840, 784, 743 697, 641; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.31 (1H, m, NH), 8.23 (2H, d, J=8.9 Hz, ArH), 7.89 (1H, s, H-2), 7.42 (2H, d, J=8.9 Hz, ArH), 5.40 (1H, d, J=6.8 Hz, CHNH), 3.86 (3H, s, OMe), 2.44 (3H, s, Me); δ_C (75 MHz; CDCl₃) 199.5, 168.0, 162.2, 154.0, 149.8 (CH), 137.9, 131.1 (CH), 130.3 (CH), 129.7, 126.7, 64.6 (CH), 55.1 (Me), 29.6 (Me); *m/z* (CI) 339/337 (M⁺, 23/100%), 333 (9), 319 (5), 307/305 (15/51), 294 (14), 277 (5), 262 (3), 234 (2), 208/206 (3/8).

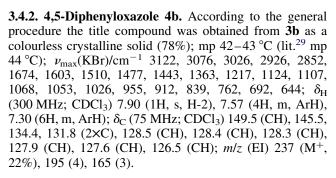
3.4. General procedure for oxazole formation

To a solution of triphenylphosphine (0.2 mmol) and iodine (0.2 mmol) in dry dichloromethane (10 ml) was added triethylamine (0.41 mmol) and then a solution of the keto amide substrate **3** in dry dichloromethane (3 ml). The reaction mixture was then stirred for 16 h and then evaporated in vacuo and purified on silica gel eluting with ethyl acetate–light petroleum (3:7) to yield the product.



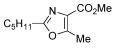
3.4.1. Methyl 5-methyloxazole-4-carboxylate 4a. According to the general procedure the title compound was obtained from **3a** as a colourless crystalline solid (45%); mp 45–47 °C (lit.²⁸ mp 46–48 °C); ν_{max} (KBr)/cm⁻¹ 3114, 3017, 2955, 2925, 2848, 1701, 1603, 1516, 1440, 1393, 1347, 1327, 1235, 1199, 1168, 1096, 1071, 968, 943, 871, 810, 779, 656; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.68 (1H, s, H-2), 3.85 (3H, s, OMe), 2.58 (3H, s, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 162.9, 157.0, 149.2 (CH), 127.5, 52.4 (Me), 12.2 (Me); *m/z* (FI) 141 (M⁺, 100%).







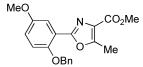
3.4.3. Methyl 5-(4-chlorophenyl)oxazole-4-carboxylate 4c. According to the general procedure the title compound was obtained from 3c as a colourless crystalline solid (65%); mp 113–115 °C (lit.³⁰ mp 111–112 °C); (Found: C, 55.5; H, 3.1; N, 5.8. C₁₁H₈CINO₃ requires C, 55.6; H, 3.4; N, 5.9%); ν_{max} (KBr)/cm⁻¹ 3126, 3014, 2960, 2924, 2854, 1705, 1618, 1524, 1489, 1441, 1373, 1325, 1250, 1209, 1097, 1072, 1005, 823, 791, 642; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.99 (2H, d, *J*=8.9 Hz, ArH) 7.84 (1H, s, H-2), 7.38 (2H, d, *J*=8.9 Hz, ArH), 3.87 (3H, s, OMe); $\delta_{\rm C}$ (100 MHz; CDCl₃) 162.3, 154.7, 149.0 (CH), 136.7,129.7 (CH), 128.9 (CH), 126.6, 125.0, 52.5 (Me); *m/z* (CI) 237/239 (M⁺, 23/13%), 234 (16), 209 (5), 208/206 (34/100), 186 (7), 139 (3).



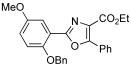
3.4.4. Methyl 5-methyl-2-pentyloxazole-4-carboxylate 4d. According to the general procedure the title compound was isolated from 3d as a colourless oil (79%); (Found: MH⁺, 212.1287. C₁₁H₁₈NO₃ requires 212.1287); $\nu_{\text{max}}(\text{film})/\text{cm}^{-1}$ 2950, 2930, 2862, 1716, 1622, 1591, 1441, 1387, 1352, 1203, 1178, 1097, 980, 825, 789, 723, 641; δ_{H} (300 MHz; CDCl₃) 3.82 (3H, s, CO₂Me), 2.66 (2H, m, CH₂), 2.52 (3H, s, Me), 1.69 (2H, m, CH₂), 1.27 (4H, m, 2×CH₂), 0.83 (3H, m, Me); δ_{C} (75 MHz; CDCl₃) 163.9,

163.7, 156.8, 127.8, 52.6 (Me), 32.1 (CH₂), 28.8 (CH₂), 27.5 (CH₂), 23.1 (CH₂), 14.7 (Me), 12.7 (Me); m/z (CI) 212 (MH⁺, 100%), 210 (8), 196 (3), 181 (5), 180 (41), 168 (6), 155 (8), 136 (2), 123 (4), 109 (1), 85 (1).

3.4.5. Ethyl 5-methyl-2-phenyloxazole-4-carboxylate 4e. According to the general procedure the title product was obtained from **3e** as a colourless solid (80%); mp 51–52 °C (ethyl acetate–light petroleum) (lit.³¹ oil); (Found: C, 67.4; H, 5.7; N, 5.9. $C_{13}H_{13}NO_3$ requires C, 67.5; H, 5.7; N, 6.1%); ν_{max} (KBr)/cm⁻¹ 3066, 2999, 2981, 2960, 2924, 2906, 2852, 1732, 1564, 1468, 1450, 1404, 1375, 1347, 1325, 1304, 1225, 1190, 1124, 1105, 1059, 1022, 841, 787, 710, 690; δ_{H} (300 MHz; CDCl₃) 8.07 (2H, m, ArH), 7.45 (3H, m, ArH), 4.45 (2H, q, *J*=7.1 Hz, OCH₂Me), 2.71 (3H, s, Me), 1.42 (3H, t, *J*=7.1 Hz, OCH₂Me); δ_{C} (75 MHz; CDCl₃) 162.9, 160.0, 156.6, 131.1 (CH), 129.2, 129.1 (CH), 127.0 (CH), 61.4 (CH₂), 14.8 (Me), 12.6 (Me), 1 Ar C unobserved; *m*/*z*(CI) 231 (M⁺, 15%), 214 (6), 187 (4), 186 (28), 185 (4).

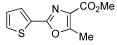


3.4.6. Methyl 2-(2-benzyloxy-5-methoxyphenyl)-5methyloxazole-4-carboxylate 4f. According to the general procedure the title compound was obtained from 3f as a white crystalline solid (23%); mp 96-99 °C (ethyl acetatelight petroleum) (lit.¹⁸ mp 98–100 °C); ν_{max} (KBr)/cm⁻¹ 3062, 3030, 3003, 2953, 2920, 2854, 2839, 1711, 1616, 1541, 1491, 1448, 1383, 1348, 1267, 1234, 1209, 1107, 1043, 868, 808, 781, 733, 692; $\delta_{\rm H}$ (400 MHz; CDCl₃) 7.54 (3H, m, ArH), 7.42 (3H, m, ArH), 6.98 (2H, m, ArH), 5.14 (2H, s, OCH₂Ph), 3.95 (3H, s, CO₂Me), 3.83 (3H, s, OMe), 2.68 (3H, s, Me); δ_C (100 MHz; CDCl₃) 175.5, 163.0, 156.4, 153.8, 151.1, 137.0, 128.4 (CH), 128.1, 127.8 (CH), 127.0 (CH), 118.7 (CH), 117.1, 115.9 (CH), 114.3 (CH), 71.7 (CH_2) , 56.0 (Me), 52.0 (Me), 12.1 (Me); m/z(EI) 353 (M⁺, 67%), 336 (19), 321 (10), 310 (16), 279 (11), 278 (25), 277 (57), 262 (100), 231 (30), 224 (53), 216 (34), 202 (22), 174 (14), 167 (22) 150 (28), 149 (65), 125 (14), 111 (23), 97 (35), 91 (83), 71 (39), 57 (52).

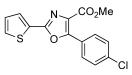


3.4.7. Ethyl 2-(2-benzyloxy-5-methoxyphenyl)-5-phenyloxazole-4-carboxylate 4g. According to the general procedure the title compound was obtained from **3g** as colourless crystalline solid (67%); mp 115–116 °C (ethyl acetate–light petroleum); (Found: C, 72.9; H, 5.4; N, 3.1. $C_{26}H_{23}NO_5$ requires C, 72.7; H, 5.4; N, 3.3%); $\nu_{max}(KBr)/$ cm⁻¹ 3053, 3027, 2996, 2971, 2925, 2863, 1711, 1588, 1537, 1491, 1419, 1373, 1224, 1107, 1035, 1015, 866, 815, 743, 687; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.71 (2H, m, ArH), 7.45

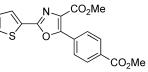
(1H, m, ArH), 7.31 (2H, m, ArH), 7.14 (6H, m, ArH), 6.83 (2H, m, ArH), 4.93 (2H, s, OCH₂Ph), 4.24 (2H, q, J=7.1 Hz, OCH₂Me), 3.66 (3H, s, OMe), 1.21 (3H, t, J=7.1 Hz, OCH₂Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 162.9, 159.2, 155.6, 154.1, 151.8, 137.1, 130.4 (CH), 128.9 (2×CH), 128.7 (CH), 128.4 (CH), 128.2 (CH), 128.1, 127.6, 119.3 (CH), 117.0, 115.6 (CH), 115.1 (CH), 72.0 (CH₂), 61.8 (CH₂), 56.4 (Me), 14.7 (Me); m/z(EI) 429 (M⁺, 34%), 412 (7), 383 (4), 355 (7), 338 (45), 324 (8), 293 (4), 279 (10), 277 (6), 224 (15), 167 (26), 151 (45), 149 (100), 139 (9), 113 (13), 105 (28), 91 (66), 71 (38), 57 (55).



3.4.8. Methyl 5-methyl-2-(thien-2-yl)oxazole-4-carboxylate **4h.** According to the general procedure the title compound was obtained from **3h** as a colourless crystalline solid (80%), 158–159 °C (diethyl ether); (Found: C, 53.8; H, 3.9; N, 6.2. $C_{10}H_9NO_3S$ requires C, 53.8; H, 4.1; N, 6.3%); $\nu_{max}(KBr)/cm^{-1}$ 3078, 3063, 2991, 2945, 2914, 2852, 1726, 1639, 1603, 1588, 1496, 1440, 1414, 1368, 1317, 1260, 1219, 1178, 1107, 1055, 1015, 984, 856, 810, 779, 769, 712, 646, 630; $\delta_H(300 \text{ MHz; CDCl}_3)$ 7.72 (1H, dd, *J*=1.1, 3.8 Hz, ArH), 7.45 (1H, dd, *J*=1.1, 5.1 Hz, ArH), 7.10 (1H, dd, *J*=3.8, 5.1 Hz, ArH), 3.93 (3H, s, OMe), 2.69 (3H, s, Me); δ_C (75 MHz; CDCl₃) 164.9, 158.1, 131.1 (CH), 131.0, 130.7 (CH), 130.5, 130.4, 130.1 (CH), 54.3 (Me), 14.3 (Me); *m/z* (EI) 223 (M⁺, 58%), 192 (23), 163 (46), 130 (100), 110 (60), 95 (77), 60 (25).

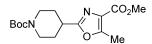


3.4.9. Methyl 5-(4-chlorophenyl)-2-(thien-2-yl)oxazole-4carboxylate 4i. According to the general procedure the title compound was obtained from 3i as a colourless crystalline solid (72%); mp 138-139 °C (methanol); (Found: C, 56.1; H, 2.9; N, 4.2. C₁₅H₁₀ClNO₃S requires C, 56.3; H, 3.2; N, 4.4%); $\nu_{\rm max}({\rm KBr})/{\rm cm}^{-1}$ 3117, 3093, 3065, 3001, 2953, 2917, 2844, 1718, 1606, 1578, 1554, 1486, 1441, 1421, 1353, 1309, 1221, 1185, 1093, 1037, 1017, 1005, 948, 924; δ_H (300 MHz; CDCl₃) 8.10 (2H, d, J=8.6 Hz, ArH), 7.82 (1H, dd, J=1.1, 3.8 Hz, ArH), 7.51 (1H, dd, J=1.1, 5.0 Hz, ArH), 7.47 (2H, d, J=8.6 Hz, ArH), 7.15 (1H, dd, J=3.8, 5.0 Hz, ArH), 3.97 (3H, s, OMe); δ_{C} (75 MHz; CDCl₃) 162.9, 156.6, 154.1, 136.9, 130.1 (2×CH), 129.7 (CH), 129.2 (CH), 128.7, 128.5 (CH), 128.4, 125.5, 53.0 (Me); m/z (EI) 321/319 (M⁺, 30/43%), 316 (12), 291 (3), 290/288 (17/45), 284 (27).



3.4.10. Methyl 5-(4-methoxycarbonylphenyl)-2-(thien-2-yl)oxazole-4-carboxylate 4j. According to the general procedure the title compound was obtained from **3j** as a colourless crystalline solid (54%); mp 168 °C (methanol);

(Found: C, 59.2; H, 3.6; N, 3.9. $C_{17}H_{13}NO_5S$ requires C, 59.5; H, 3.8; N, 4.1%); ν_{max} (KBr)/cm⁻¹ 3109, 3088, 3032, 2996, 2950, 2843, 1716, 1609, 1578, 1501, 1434, 1404, 1347, 1281, 1224, 1189, 1107, 1091, 1020, 1015, 943, 861, 810, 774, 723, 692; δ_{H} (300 MHz; DMSO) 8.26 (2H, d, *J*=8.7 Hz, ArH), 8.13 (2H, d, *J*=8.7 Hz, ArH), 7.96 (2H, m, ArH), 7.32 (1H, app t, *J*=4.1 Hz, ArH), 3.93 (3H, s, OMe), 3.90 (3H, s, OMe); δ_{C} (75 MHz; DMSO) 165.9, 162.0, 156.3, 152.8, 131.7 (CH), 131.1, 130.8, 130.3 (CH), 129.6 (CH), 129.2, 129.1 (CH), 128.7 (CH), 128.0, 52.8 (Me), 52.6 (Me); *m/z* (CI) 343 (M⁺, 33%), 340 (5), 314 (3), 313 (6), 312 (29).



3.4.11. Methyl 2-(1-tert-butoxycarbonylpiperidin-4-yl)-5-methyloxazole-4-carboxylate 4k. According to the general procedure the title compound was isolated from **3k** as a colourless crystalline solid (66%); mp 101–103 °C; (Found: C, 59.1; H, 7.6; N, 8.5. C₁₆H₂₄N₂O₅ requires C, 59.2; H, 7.5; N, 8.6%); ν_{max}(KBr)/cm⁻¹ 2981, 2962, 2937, 2850, 1718, 1682, 1622, 1425, 1346, 1252, 1234, 1211, 1167, 1107, 1026, 937, 783; $\delta_{\rm H}$ (300 MHz; CDCl₃) 4.09 (2H, m, 2×CH), 3.87 (3H, s, CO₂Me), 2.90 (3H, m, 3×CH), 2.58 (3H, s, Me), 1.97 (2H, m, 2×CH), 1.75 (2H, m, 2×CH), 1.43 (9H, s, CMe₃); δ_C (75 MHz; CDCl₃) 164.8, 163.2, 156.5, 155.0, 127.5, 80.0, 52.3 (Me), 43.5 (CH), 36.0 (CH), 29.7 (CH₂), 28.8 (Me), 12.3 (Me); *m/z* (CI) 324 (M⁺, 2%), 298 (3), 297 (11), 271 (3), 270 (15), 269 (100), 251 (55), 226 (4), 225 (25), 223 (32), 193 (3), 168 (14), 155 (2), 136 (5), 83 (5).

3.5. General procedure for thiazole formation

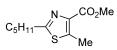
A solution of the keto amide substrate **3** (1.5 mmol) and Lawesson's reagent (3.0 mmol) in dry THF (10 ml) was heated to reflux for 4-6 h. The reaction mixture was then evaporated in vacuo and purified on silica gel eluting with ethyl acetate–light petroleum (3:7) to yield the product.

$$\langle \mathbf{x} | \mathbf{x}$$
 \mathbf{x} $\mathbf{x$

3.5.1. Methyl 5-methylthiazole-4-carboxylate 5a. According to the general procedure the title compound was obtained from **3a** as a colourless crystalline solid (60%); mp 62–65 °C (diethyl ether) (lit.³² mp not given); (Found: M⁺, 157.0200. C₆H₇NO₂S requires 157.0198); $\nu_{\rm max}$ (KBr)/cm⁻¹ 3035, 2958, 2924, 2850, 1722, 1597, 1518, 1433, 1372, 1333, 1288, 1203, 1124, 1066, 955, 881, 831, 785, 762, 627; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.51 (1H, s, H-2), 3.87 (3H, s, OMe), 2.73 (3H, s, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 163.2, 149.6 (CH), 145.4, 142.1, 52.6 (Me), 13.4 (Me); *m/z* (EI) 157 (M⁺, 17%), 127 (6), 126 (65), 125 (100), 98 (12), 97 (23), 72 (4), 71 (11), 59 (8), 54 (3).

3.5.2. 4,5-Diphenylthiazole 5b. According to the general

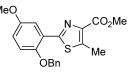
procedure the title compound was obtained from **3b** as a colourless crystalline solid (94%); mp 59–61 °C (lit.³³ mp 60–61 °C); (Found: C, 75.7; H, 4.6; N, 5.8. C₁₅H₁₁NS requires C, 75.9; H, 4.7; N, 5.9%); ν_{max} (KBr)/cm⁻¹ 3053, 2926, 2854, 2808, 1497, 1475, 1441, 1414, 1338, 1279, 1070, 1026, 999, 966, 899, 825, 766, 694; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.84 (1H, s, H-2), 7.54 (2H, m, ArH), 7.34 (8H, m, ArH); $\delta_{\rm C}$ (75 MHz; CDCl₃) 151.5 (CH), 151.0, 135.0, 133.4, 132.2, 130.1 (CH), 129.4 (CH), 129.2 (CH), 128.73, 128.71, 128.3 (CH); *m/z* (EI) 237 (M⁺, 24%), 236 (6).



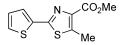
3.5.3. Methyl 5-methyl-2-pentylthiazole-4-carboxylate 5d. According to the general procedure the title compound was isolated from 3d as a colourless oil (89%); (Found: MH⁺, 228.1050. C₁₁H₁₈NO₂S requires 228.1058); $\nu_{max}(film)/cm^{-1}$ 2954, 2929, 2858, 1716, 1504, 1437, 1221, 1068, 962, 866, 789, 768; $\delta_{\rm H}$ (300 MHz; CDCl₃) 3.88 (3H, s, OMe), 2.91 (2H, t, *J*=7.8 Hz, CH₂), 2.69 (3H, s, Me), 1.70 (2H, m, CH₂), 1.34 (4H, m, 2×CH₂), 0.85 (3H, t, *J*=7.0 Hz, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 168.2, 163.3, 144.9, 140.5, 52.4 (Me), 33.8 (CH₂), 31.6 (CH₂), 30.2 (CH₂), 22.7 (CH₂), 14.3 (Me), 13.4 (Me); *m/z* (CI) 228 (MH⁺, 100%), 226 (8), 212 (3), 197 (4), 196 (31), 184 (3), 171 (11), 139 (3).



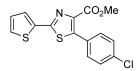
3.5.4. Ethyl 5-methyl-2-phenylthiazole-4-carboxylate 5e. According to the general procedure the title product was obtained from **3e** as a colourless solid (53%); mp 80–81 °C (ethyl acetate–light petroleum) (lit.³⁴ mp 59–61 °C); (Found: C, 63.1; H, 5.2; N, 5.6. $C_{13}H_{13}NO_2S$ requires C, 63.1; H, 5.3; N, 5.7%); $\nu_{max}(KBr)/cm^{-1}$ 2980, 2927, 2902, 2868, 2852, 1707, 1517, 1466, 1443, 1367, 1327, 1242, 1219, 1165, 1065, 1018, 974, 777, 698, 638; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.15 (2H, m, ArH), 7.65 (3H, m, ArH), 4.66 (2H, q, *J*=7.1 Hz, OCH₂Me), 3.03 (3H, s, Me), 1.67 (3H, t, *J*=7.1 Hz, OCH₂Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 164.1, 163.1, 144.9, 142.7, 133.3, 130.7 (CH), 129.3 (CH), 127.1 (CH), 61.6 (CH₂), 14.8 (Me), 13.8 (Me); *m/z*(CI) 247 (M⁺, 13%), 230 (5), 203 (4), 202 (30), 201 (4).



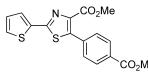
3.5.5. Methyl 2-(2-benzyloxy-5-methoxyphenyl)-5methylthiazole-4-carboxylate 5f. According to the general procedure the title product was obtained from 3f as a colourless crystalline solid (55%); mp 115–117 °C (methanol); (Found: C, 64.9; H, 5.0; N, 3.6. $C_{20}H_{19}NO_4S$ requires C, 65.0; H, 5.2; N, 3.8%); $\nu_{max}(KBr)/cm^{-1}$ 3058, 3032, 2996, 2946, 2914, 2868, 2838, 1701, 1650, 1609, 1506, 1455, 1440, 1414, 1388, 1312, 1276, 1235, 1173, 1117, 1071, 1035, 999, 871, 805, 779, 733, 692, 666; δ_{H} (400 MHz; CDCl₃) 7.94 (1H, d, *J*=3.1 Hz, ArH), 7.41 (5H, m, ArH), 6.95 (2H, m, ArH), 5.23 (2H, s, OCH₂Ph), 3.96 (3H, s, OMe), 3.87 (3H, s, OMe), 2.75 (3H, s, Me); δ_{C} (100 MHz; CDCl₃) 163.4, 158.1, 154.0, 149.9, 145.6, 139.8, 136.3, 128.6 (CH), 128.2 (CH), 127.8 (CH), 122.7, 117.9 (CH), 114.4 (CH), 112.0 (CH), 71.6 (CH₂), 56.0 (Me), 52.0 (Me), 12.9 (Me); *m*/*z*(EI) 369 (M⁺, 8%), 352 (7), 279 (5), 278 (16), 246 (5), 218 (8), 205 (7), 177 (7), 149 (100), 125 (5), 111 (8), 97 (12), 91 (23), 83 (12), 55 (16).



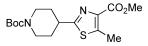
3.5.6. Methyl **5-methyl-2-(thien-2-yl)thiazole-4-carboxylate 5h.** According to the general procedure the title product was obtained from **3h** as a colourless crystalline solid (69%); mp 113–115 °C (methanol); (Found: C, 49.9; H, 3.6; N, 5.7. C₁₀H₉NO₂S₂ requires C, 50.2; H, 3.8; N, 5.9%); ν_{max} (KBr)/cm⁻¹ 3109, 2991, 2950, 2919, 2843, 1711, 1521, 1470, 1440, 1419, 1378, 1317, 1240, 1219, 1163, 1071, 912, 856, 840, 825, 781, 764, 702, 630; $\delta_{\rm H}$ (300 MHz; CDCl₃) 7.47 (1H, d, *J*=3.7 Hz, ArH), 7.40 (1H, d, *J*=5.0 Hz, ArH), 7.06 (1H, dd, *J*=3.7, 5.0 Hz, ArH), 3.94 (3H, s, OMe), 2.78 (3H, s, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 163.2, 158.1, 144.8, 141.7, 136.8, 128.5 (CH), 128.2 (CH), 127.6 (CH), 52.6 (Me), 13.6 (Me); *m*/*z*(EI) 239 (M⁺, 64%), 209 (8), 208 (21), 207 (65), 181 (8), 179 (47), 136 (4), 129 (9), 127 (100), 111 (5), 99 (4), 71 (8), 59 (10).



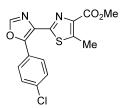
3.5.7. Methyl 5-(4-chlorophenyl)-2-(thien-2-yl)thiazole-4-carboxylate 5i. According to the general procedure the title compound was obtained from 3i as a pale crystalline solid (89%); mp 146–149 °C (methanol); (Found: C, 53.2; H, 2.8; N, 4.0. $C_{15}H_{10}CINO_2S_2$ requires C, 53.6; H, 3.0; N, 4.2%); $\nu_{max}(KBr)/cm^{-1}$ 3109, 3088, 3068, 3049, 3032, 2996, 2951, 2924, 1716, 1647, 1541, 1466, 1431, 1417, 1400, 1335, 1201, 1169, 1088, 1016, 999; δ_{H} (300 MHz; CDCl₃) 7.55 (1H, dd, *J*=1.1, 3.7 Hz, ArH), 7.46 (3H, m, ArH), 7.41 (2H, d, *J*=8.7 Hz, ArH), 7.10 (1H, dd, *J*=3.7, 5.0 Hz, ArH), 3.86 (3H, s, CO₂Me); δ_{C} (75 MHz; CDCl₃) 163.0, 160.8, 145.1, 141.2, 136.7, 136.3, 132.0 (CH), 129.6, 129.3 (CH), 129.1 (CH), 128.8 (CH), 128.5 (CH), 52.9 (Me); *m*/*z* (CI) 335/337 (M⁺, 41/34%), 332 (12), 307 (5), 306/304 (27/60), 300 (7).



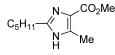
3.5.8. Methyl 5-(4-methoxycarbonylphenyl)-2-(thien-2yl)thiazole-4-carboxylate 5j. According to the general procedure the title compound was obtained from 3j as a pale crystalline solid (34%); mp 165–167 °C (methanol); (Found: C, 56.6; H, 3.5; N, 3.7. $C_{17}H_{13}NO_4S_2$ requires C, 56.8; H, 3.7; N, 3.9%); $\nu_{max}(KBr)/cm^{-1}$ 3119, 3117, 2955, 2914, 2848, 1721, 1654, 1603, 1537, 1470, 1424, 1327, 1281, 1260, 1214, 1178, 1112, 1081, 963, 917, 851, 764, 702; δ_H (300 MHz; DMSO) 8.06 (2H, d, *J*=8.5 Hz, ArH), 7.87 (1H, dd, *J*=1.1, 4.9 Hz, ArH), 7.82 (1H, dd, *J*=1.1, 3.8 Hz, ArH), 7.75 (2H, d, J=8.5 Hz, ArH), 7.25 (1H, dd, J=3.8, 4.9 Hz, ArH), 3.92 (3H, s, OMe), 3.78 (3H, s, OMe); $\delta_{\rm C}$ (75 MHz; DMSO) 166.1, 162.0, 160.0, 143.3, 141.0, 135.5, 134.6, 130.7 (CH), 130.5 (CH), 130.4, 129.5 (CH), 129.3 (CH), 129.1 (CH), 52.7 (Me), 52.5 (Me); m/z (CI) 359 (M⁺, 30%), 356 (3), 329 (7), 328 (32), 285 (3).



3.5.9. Methyl 2-(1-tert-butoxycarbonylpiperidin-4-yl)-5methylthiazole-4-carboxylate 5k. According to the general procedure the title compound was isolated from **3k** as a colourless crystalline solid (74%); mp 75–77 °C; (Found: M⁺, 340.1439. C₁₆H₂₄N₂O₄S requires 340.1457); $\nu_{\rm max}$ (KBr)/cm⁻¹ 3174, 2980, 2958, 2933, 2872, 1741, 1709, 1597, 1454, 1377, 1338, 1306, 1265, 1238, 1205, 1155, 1111, 1088, 1053, 933, 856, 746; $\delta_{\rm H}$ (300 MHz; CDCl₃) 4.20 (2H, m, 2×CH), 3.90 (3H, s, OMe), 3.16 (1H, m, CH), 2.79 (2H, m, 2×CH), 2.73 (3H, s, Me), 2.04 (2H, m, 2×CH), 1.63 (2H, m, 2×CH), 1.44 (9H, s, CMe₃); δ_{C} (75 MHz; CDCl₃) 168.9, 161.0, 152.7, 142.4, 138.3, 77.8, 50.2 (Me), 41.6 (CH₂), 39.0 (CH), 30.5 (CH₂), 26.5 (Me), 11.3 (Me); m/ z (CI) 340 (M⁺, 3%), 314 (3), 313 (14), 287 (5), 286 (9), 285 (100), 267 (40), 253 (4), 242 (9), 241 (56), 239 (33), 210 (2), 209 (7), 197 (3), 184 (24), 171 (3), 152 (9), 83 (7).



3.5.10. Methyl 2-[5-(4-chlorophenyl)oxazol-4-yl]-5methylthiazole-4-carboxylate 5I. According to the general procedure 3I was treated with Lawesson's reagent to yield the title compound as a colourless crystalline solid (40%); mp 210–212 °C; (Found: C, 53.6; H, 3.0; N, 8.2. C₁₅H₁₁-ClN₂O₃S requires C, 53.8; H, 3.3; N, 8.4%); ν_{max} (KBr)/ cm⁻¹ 3129, 3058, 3037, 2955, 2909, 2843, 1701, 1511, 1470, 1434, 1327, 1230, 1112, 1091, 1066, 1030, 1009, 912, 876, 839, 769, 744, 625; $\delta_{\rm H}$ (300 MHz; CDCl₃) 8.47 (2H, d, *J*=8.7 Hz, ArH), 7.88 (1H, s, H-2), 7.42 (2H, d, *J*=8.7 Hz, ArH), 3.93 (3H, s, OMe), 2.79 (3H, s, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 160.9, 154.6, 147.2 (CH), 145.0, 144.2, 139.6, 133.7, 127.2, 126.8 (CH), 126.7 (CH), 123.8, 50.2 (Me), 11.1 (Me); *m/z* (FI) 334 (M⁺, 100%).



3.5.11. Methyl 5-methyl-2-pentyl-1*H*-imidazole-4-carboxylate 6a. Ketoamide 3a (1.0 g, 4.4 mmol), ammonium acetate (510 mg, 6.6 mmol) and glacial acetic acid (2 ml) were dissolved in toluene (50 ml) and heated under reflux for 2 h. The reaction was cooled, washed with brine, dried (MgSO₄), reduced in vacuo and purified by column chromatography to yield the title product as a colourless oil (756 mg, 82%); (Found: MH⁺, 211.1454. C₁₁H₁₉N₂O₂ requires 211.1447); ν_{max} (film)/cm⁻¹ 3399, 3308, 2956,

2928, 2866, 1739, 1602, 1539, 1434, 1373, 1292, 1264, 1187, 1106; $\delta_{\rm H}$ (300 MHz; CDCl₃) 6.30 (1H, s, NH), 3.63 (3H, s, OMe), 2.24 (2H, t, *J*=7.4 Hz, CH₂), 1.91 (3H, s, Me), 1.71–1.61 (2H, m, CH₂), 1.35–1.29 (4H, m, 2×CH₂), 0.88 (3H, t, *J*=6.8 Hz, Me); $\delta_{\rm C}$ (75 MHz; CDCl₃) 174.5, 171.6, 168.4, 93.3, 51.0 (Me), 36.8 (CH₂), 31.7 (CH₂), 25.9 (CH₂), 22.8 (CH₂), 19.5 (Me), 14.2 (Me); *m/z* (CI) 211 (MH⁺, 28%), 198 (12), 197 (100), 195 (8), 179 (5), 157 (5), 154 (3), 131 (6), 130 (46), 129 (13), 113 (3), 99 (5), 98 (2), 70 (3).

$$C_5H_{11} \xrightarrow{N} CO_2Me$$

 $N Me$

3.5.12. Methyl 1,5-dimethyl-2-pentyl-1H-imidazole-4carboxylate 6b. Ketoamide 3a (250 mg, 1.1 mmol) and glacial acetic acid (2 ml) were dissolved in toluene (50 ml) to which was then added a solution of methylamine (2 M in THF, 0.82 ml, 1.7 mmol) and heated under reflux for 2 h. The reaction was cooled, washed with brine, dried (MgSO₄), reduced in vacuo and purified by column chromatography to yield the title product as an orange oil (114 mg, 46%); (Found: MH⁺, 225.1603. C₁₂H₂₁N₂O₆ requires 225.1603); $\nu_{\text{max}}(\text{film})/\text{cm}^{-1}$ 2958, 2928, 2858, 1701, 1577, 1530, 1439, 1373, 1216, 1068; δ_H (300 MHz; CDCl₃) 3.81 (3H, s, OMe), 3.42 (3H, s, NMe), 2.63 (2H, t, J=7.7 Hz, CH₂), 2.46 (3H, s, Me), 1.65–1.62 (2H, m, CH₂), 1.30-1.28 (4H, m, 2×CH₂), 0.84 (3H, t, J=6.6 Hz, Me); δ_{C} (75 MHz; CDCl₃) 164.7, 148.7, 136.5, 127.1, 51.5 (Me), 31.9 (CH₂), 30.5 (Me), 27.9 (CH₂), 27.6 (CH₂), 22.6 (CH₂), 14.2 (Me), 10.4 (Me); *m/z* (CI) 225 (MH⁺, 100%), 224 (8), 209 (3), 195 (3), 194 (6), 193 (56), 181 (6), 169 (2), 168 (14), 150 (2), 136 (5), 116 (2), 99 (3).

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References and notes

- Part 4, Buck, R. T.; Moody, C. J.; Pepper, A. G. ARKIVOC 2002, 16–32, http://arkat-usa.org/ark/journal/2002/Padwa/ AP-391H/391H.htm.
- Roy, R. S.; Gehring, A. M.; Milne, J. C.; Belshaw, P. J.; Walsh, C. T. *Nat. Prod. Rep.* **1999**, *16*, 249–263.
- Schwarzer, D.; Finking, R.; Marahiel, M. A. Nat. Prod. Rep. 2003, 20, 275–287.
- Gordon, T. D.; Singh, J.; Hansen, P. E.; Morgan, B. A. *Tetrahedron Lett.* **1993**, *34*, 1901–1904.
- 5. Falorni, M.; Dettori, G.; Giacomelli, G. *Tetrahedron:* Asymmetry **1998**, *9*, 1419–1426.
- Christensen, C.; Schiodt, C. B.; Foged, N. T.; Meldal, M. QSAR Comb. Sci. 2003, 22, 754–766.
- Palmer, D. C.; Venkatraman, S. Synthesis and reactions of oxazoles. In Oxazoles: synthesis, reactivity and spectroscopy.

Part A; Palmer, D. C., Ed.; Wiley: Hoboken, New Jersey, 2003; pp 1–390.

- Singh, J.; Gordon, T. D.; Earley, W. G.; Morgan, B. A. *Tetrahedron Lett.* 1993, 34, 211–214.
- 9. Wipf, P.; Miller, C. P. J. Org. Chem. 1993, 58, 3604-3606.
- Bagley, M. C.; Buck, R. T.; Hind, S. L.; Moody, C. J. J. Chem. Soc., Perkin Trans. 1 1998, 591–600.
- Moody, C. J.; Bagley, M. C. J. Chem. Soc., Perkin Trans. 1 1998, 601–607.
- Bagley, M. C.; Bashford, K. E.; Hesketh, C. L.; Moody, C. J. J. Am. Chem. Soc. 2000, 122, 3301–3313.
- 13. Clapham, B.; Spanka, C.; Janda, K. D. Org. Lett. 2001, 3, 2173–2176.
- Clapham, B.; Lee, S. H.; Koch, G.; Zimmermann, J.; Janda, K. D. *Tetrahedron Lett.* **2002**, *43*, 5407–5410.
- 15. Regitz, M.; Maas, G. Diazo compounds. Properties and synthesis; Academic: Orlando, Florida, 1986.
- Baum, J. S.; Shook, D. A.; Davies, H. M. L.; Smith, H. D. Synth. Commun. 1987, 17, 1709–1716.
- Morrison, H.; Danishefsky, S.; Yates, P. J. Org. Chem. 1961, 26, 2617–2618.
- Yokokawa, F.; Izumi, K.; Omata, J.; Shioiri, T. *Tetrahedron* 2000, 56, 3027–3034.
- For a review on Lawesson's reagent, see: Jesberger, M.; Davis, T. P.; Barner, L. Synthesis 2003, 1929–1958.
- For other examples of thiazole formation from ketoamides, see Ref. 4, and Buchanan, J. L.; Mani, U. N.; Plake, H. R.; Holt, D. A. *Tetrahedron Lett.* **1999**, *40*, 3985–3988; Frantz, D. E.; Morency, L.; Soheili, A.; Murry, J. A.; Grabowski, E. J. J.; Tillyer, R. D. *Org. Lett.* **2004**, *6*, 843–846.
- Schlewer, G.; Wermuth, C. G.; Chambon, J. P. Eur. J. Med. Chem. 1984, 19, 181–186.
- McGuiness, M.; Shechter, H. Tetrahedron Lett. 1990, 31, 4987–4990.
- 23. Looker, J. H.; Hayes, C. H. J. Org. Chem. 1963, 28, 1342-1347.
- 24. Bethell, D.; Parker, V. D. J. Am. Chem. Soc. 1986, 108, 7194–7200.
- 25. Davidson, D.; Weiss, M.; Jelling, M. J. Org. Chem. 1937, 2, 328-334.
- 26. Nunami, K.; Hiramatsu, K.; Hayashi, K.; Matsumoto, K. *Tetrahedron* **1988**, *44*, 5467–5478.
- 27. Bratusek, U.; Hvala, A.; Stanovnik, B. J. Heterocycl. Chem. **1998**, *35*, 1281–1284.
- Suzuki, M.; Iwasaki, T.; Miyoshi, M.; Okumura, K.; Matsumoto, K. J. Org. Chem. 1973, 38, 3571–3575.
- Maeda, M.; Kojima, M. J. Chem. Soc., Perkin Trans. 1 1977, 239–247.
- Ozaki, Y.; Maeda, S.; Iwasaki, T.; Matsumoto, K.; Odawara, A.; Sasaki, Y.; Morita, T. *Chem. Pharm. Bull.* **1983**, *31*, 4417–4424.
- Dietliker, K.; Gilgen, P.; Heimgartner, H.; Schmid, H. *Helv. Chim. Acta* 1976, *59*, 2074–2099.
- Ernest, I.; Gosteli, J.; Greengrass, C. W.; Holick, W.; Jackman, D. E.; Pfaendler, H. R.; Woodward, R. B. J. Am. Chem. Soc. 1978, 100, 8214–8222.
- Maeda, M.; Kojima, M. J. Chem. Soc., Perkin Trans. 1 1978, 685–692.
- Cornwall, P.; Dell, C. P.; Knight, D. W. J. Chem. Soc., Perkin Trans. 1 1991, 2417–2428.